



UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS International General Certificate of Secondary Education

NUMBER	NUMBER

1 hour

Candidates answer on the Question Paper.

No Additional Materials are required.

READ THESE INSTRUCTIONS FIRST

Write your Centre number, candidate number and name on all the work you hand in.

Write in dark blue or black pen.

You may use a pencil for any diagrams or graphs.

Do not use staples, paper clips, highlighters, glue or correction fluid.

DO NOT WRITE IN ANY BARCODES.

Answer all questions.

At the end of the examination, fasten all your work securely together.

The number of marks is given in brackets [] at the end of each question or part question.

For Examiner's Use				
1				
2				
3				
Total				

This document consists of 8 printed pages and 4 blank pages.

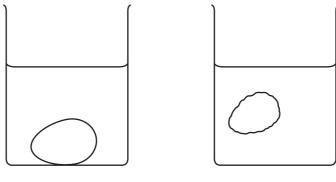


1 Two uncooked hen's eggs of similar size were submerged in acid to dissolve the shell. The contents of the eggs are left intact, enclosed by the egg membrane.

One de-shelled egg has been placed in distilled water for 2 days. The other egg has been placed in salt solution for 2 days.

For Examiner's Use

Fig. 1.1 shows the two eggs after two days.



distilled water

salt solution

Fig. 1.1

(a) (i) Complete Table 1.1 to compare the two eggs.

Та	bl	е	1		1
		•		•	

	egg in water	egg in salt solution
size of the egg		
position in the liquid		
external appearance of the egg		

[3]

(ii) Suggest an explanation for these differences.

[5]

3

Question 1 continues on page 4

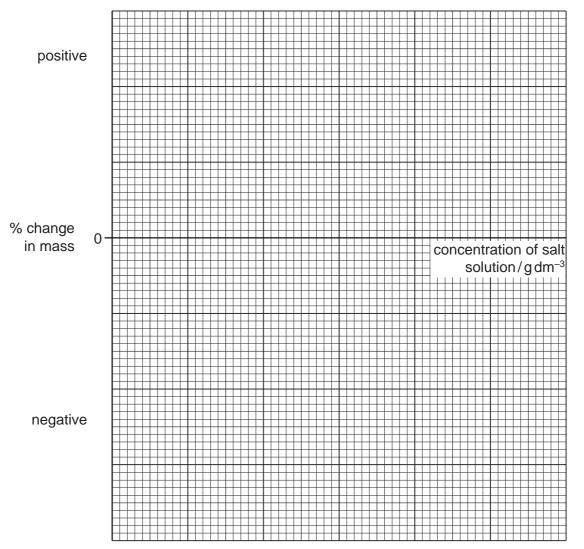
(b) A student carried out a similar experiment and varied the concentration of salt in different solutions.

Table 1.2 shows the percentage change in mass of the eggs after 2 days.

% change in mass
+7.3
+3.2
+0.8
-2.5
-6.4
-10.8

Table 1.2

(i) Draw a graph of this data on the axes.



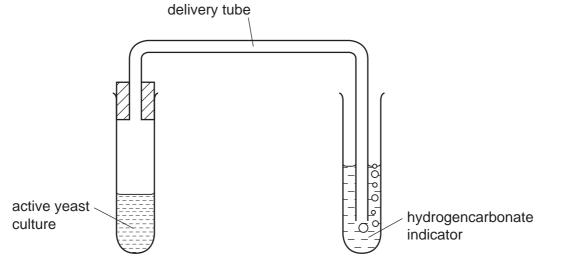
(ii) Determine the concentration of salt solution where there is no change in mass. [1] (iii) Explain why there is no change in mass at this concentration of salt solution. [3] (c) Describe a food test which might show whether there is more protein to be found in the 'white' or in the 'yolk' of the egg. [4] [Total: 19]

For Examiner's Use

6

2 An active yeast culture is placed in a test-tube. This test-tube is connected by a delivery tube to a second test-tube containing hydrogencarbonate indicator shown in Fig. 2.1.

Hydrogencarbonate indicator changes colour according to pH. In alkaline solutions the indicator becomes purple and in acidic solutions the indicator becomes yellow.





At the start of the experiment the indicator is red in colour. After 15 minutes the indicator becomes yellow.

(a) Explain the reason for this colour change.

[3]

For

Examiner's Use (b) Describe how you would collect and measure the volume of gas given off per minute by 20 cm³ of the active yeast culture.

For Examiner's Use

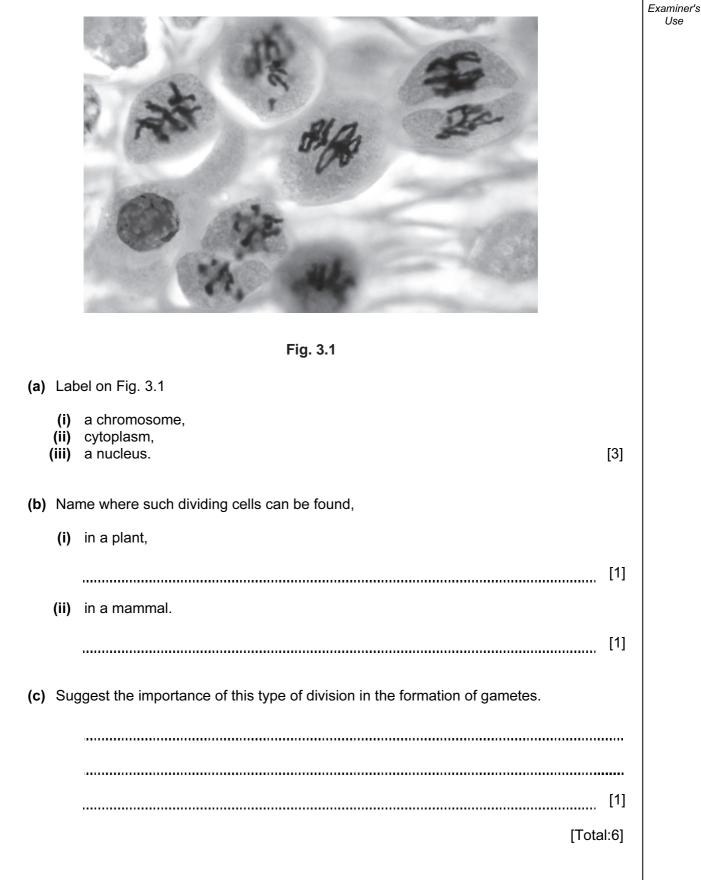
[6] (c) Fig. 2.2 shows a yeast cell reproducing asexually. Х ×5000



(i) Make a labelled drawing of the yeast cells.

For Examiner's Use

(ii)	Measure the diameter of the yeast cell in Fig. 2.2 between X and Y .	[3]
	diameter (distance between X and Y)	
	Calculate the magnification of your drawing. Show your working.	
	magnification	[3]
	[Total:	15]



3 Fig. 3.1 shows cells dividing to form gametes.

For

Use

Permission to reproduce items where third-party owned material protected by copyright is included has been sought and cleared where possible. Every reasonable effort has been made by the publisher (UCLES) to trace copyright holders, but if any items requiring clearance have unwittingly been included, the publisher will be pleased to make amends at the earliest possible opportunity.

University of Cambridge International Examinations is part of the Cambridge Assessment Group. Cambridge Assessment is the brand name of University of Cambridge Local Examinations Syndicate (UCLES), which is itself a department of the University of Cambridge.