NO:	FORMULA OF COMPOUNDS-SET-1				
1	For which compound is the formula correct?				
		compound	formula	a	
	A	ammonia	$NH_4$		
	В	carbon monoxide	$CO_2$		
	С	iron(III) oxide	Fe <sub>3</sub> O <sub>2</sub>		
	D	zinc hydroxide	Zn(OH)	)2	
Ms-1	D				
2	For which compound is the formula correct?				
		compound	formula		
	Α	ammonia	NH <sub>4</sub>		
	В	carbon dioxide	СО		
	С	potassium oxide	P <sub>2</sub> O		
	D	zinc chloride	ZnCl <sub>2</sub>		
N4- 2					
Ms-2	D				

The electronic structures of atoms P and Q are shown. key = electron P and Q react to form an ionic compound. What is the formula of this compound?  $\mathbf{B} \quad \mathsf{P}_2\mathsf{Q}$  $\mathbf{C}$   $P_2Q_6$  $\mathbf{A}$  PQ<sub>2</sub>  $\mathbf{D} \quad \mathsf{P}_6\mathsf{Q}_2$ Ms-3 Α For which compound is the formula correct? formula compound ammonium chloride Α NH<sub>3</sub>C1 CuS copper(II) sulphide В iron(II) sulphide Fe<sub>3</sub>S C silver nitrate  $Ag_2NO_3$ D Ms-4

5	The diagram shows a molecule of vinyl chloride (used to make pvc).				
	key  a carbon atom a chlorine atom a hydrogen atom  What is the formula of vinyl chloride?				
	<b>A</b> $CH_2Cl_3$ <b>B</b> $CH_3Cl_2$ <b>C</b> $C_2HCl_3$ <b>D</b> $C_2H_3Cl$				
Ms-5	D				
6	When written as formulae, which compound has the greatest number of oxygen atoms?				
	A calcium oxide				
	B copper(II) oxide				
	C iron(III) oxide				
	<b>D</b> potassium oxide				
Ms-6	С				

The electronic structures of atoms P and Q are shown. key = electron P and Q react to form an ionic compound. What is the formula of this compound?  $PQ_2$  $\mathbf{B} \quad \mathsf{P}_2\mathsf{Q} \qquad \qquad \mathbf{C} \quad \mathsf{P}_2\mathsf{Q}_6$  $\mathbf{D} \quad \mathsf{P}_6\mathsf{Q}_2$ Ms-7 8 A compound contains one atom of calcium, two atoms of hydrogen and two atoms of oxygen. What is the correct chemical formula of the compound? C H<sub>2</sub>CaO<sub>2</sub>  $\mathbf{A}$  CaO<sub>2</sub>H<sub>2</sub> **B** HOCaOH D Ca(OH)<sub>2</sub> D Ms-8