MARK SCHEME for the May/June 2012 question paper

for the guidance of teachers

0653 COMBINED SCIENCE

0653/23

Paper 2 (Core Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

• Cambridge will not enter into discussions or correspondence in connection with these mark schemes.

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	Page 2	2	Mark Scheme: Teachers' version	Syllabus	Paper	
			IGCSE – May/June 2012	0653	23	
1			=) distance / time ; = 12.5 (km / h) ;		[2]	
	(b) (i)	cher	mical ;		[1]	
	(ii)	(ii) heating engine / heating surroundings / light / sound ;				
	me	etal ca	ack expands in summer / hot weather ; in expand into gap ; s damage to tracks ;		[max 2]	
	pre	events			[max 2]	
	(d) eth	(d) ethanol is renewable source / no sulfur dioxide produced ;				
	(e) (i) 5 (km		m/h); (accept 4 / reference to greater than 3 but eq	ual to and less than {	5) [1]	
	(ii)	1200	0 (W) ;		[1]	
	(iii)	wind	d speed variable / wind unreliable / owtte ;		[1]	
					[Total: 10]	
2	(a) (i)	nucl	eus ;		[1]	
	(ii)	18 ; evid	ence of neutrons = nucleon number minus proton n	umber ;	[2]	
	(iii)	hydr	rogen ;		[1]	
	(b) (i)	e.g.	typical metal property for X and corresponding non- X conductor Y insulator X malleable Y not malleable	metal for Y ;	[1]	
	(ii)	Y is	reactive / specific example, ${f Z}$ is unreactive ;		[1]	
	(c) (i)		rence to oxygen (not air) ; ch joins / reacts with carbon ;		[2]	
	(ii)		(too) acidic ;			
			e, reacts with / neutralises acid / reduces acidity ; acrease fertility ;		[max 2]	
					[Total: 10]	

Pa	Page 3		Mark Scheme: Teachers' version	Syllabus	Paper
			IGCSE – May/June 2012 0653		23
3 (a)	only eats		al that eats plants ; s plants / does not eat meat ; e to getting energy from, its food / plants ;		[max 2]
(b)	eat	eat a lot ; eat / take in, more energy than they use ; excess, carbohydrate / protein, converted to fat ;			
(c)	(i)	 (i) the greater the body mass, the greater the chance of survival; idea that effect is greater at lower body masses / levels off at higher body masses; use of figures; 			
	(ii)		lator / poor conductor / reduces conduction ;		[max 2] [1]
(d)	carbon dioxide ; methane ;				[2]
(e)	(i)				[1]
	(ii)				[max 1] [Total: 11]
4 (a)	insert b measu		nesium to acid ; ng ; time for known volume of gas to collect / measure v rvals / measure how long reaction lasts ;	olume of gas at fixed	[3]
(b)	(i)	temp	perature / surface area of magnesium ;		[1]
	(ii)		for (same volume) of gas to collect is shorter ; ause rate of reaction is greater ;		[2]
(c)	(i)	atom	s charged, atom is neutral / proton and electron num n, but differ in the ion / the atom is reactive, the ion is tron shells full, while atom outer shell is not full ;		[1]
	(ii)	MgC	Cl ₂ ;		[1]
	(iii)	(iii) hydrogen ;			[1]
					[Total: 9]

	Page 4		•	Mark Scheme: Teachers' version	Syllabus	Paper		
				IGCSE – May/June 2012	0653	23		
5	(a)	(i)	(i) between 10 and 20 (Hz) to between 20 000 and 25 000 (Hz) ;					
		 (ii) frequency – number of waves produced / passing a point per second ; wavelength – distance between consecutive, peaks / troughs ; 						
	(b)	(i)	need	ds, air / particles / a medium ;		[1]		
		(ii) quieter / lower volume ;						
		(iii)	micr	rowaves ;		[1]		
	(c)			incidence labelled ; refraction labelled ;		[2] [Total: 8]		
6	(a)	labe	el to r	root hair cell ;		[1]		
	(b)	 (b) water ; minerals / ions / named mineral ion ; (allow two different mineral ions for 2 marks) 						
	(c)	(i)	xylei	m ;		[1]		
		(ii)	A in	central area of root ;		[1]		

(d) (i)

structure	plant cells	animal cells
cell membrane	\checkmark	\checkmark
cell wall	\checkmark	×
nucleus	\checkmark	\checkmark
vacuole containing sap	\checkmark	×

1 mark for any row correct ;;;;

[4]

[1]

(ii) no, as they do not get light / as they are underground ;

[Total: 10]

	Page 5				Syllabus	
				IGCSE – May/June 2012	0653	23
7	(a)	(i)	voltmeter in	series with lamp ; parallel with lamp ; istor in series with lamp and all else corre	ect;	[3]
		(ii)	to change, voltage across / current through, lamp / in circuit ;			[1]
	(b)	(i)	positive and	I negative ;		[1]
		(ii)	electrons ;			[1]
	(iii)		metal / nam	ed metal / graphite ;		[1]
						[Total: 7]
8	8 (a) (i)) nitrogen 78% oxygen 21% others 1% ;; (all correct 2, two correct 1)			
		(ii)	carbon dioxide / water (vapour) / any noble gas ;			
	 (b) (i) elements all atoms have same proton number / are found in Periodic Table / car broken down into anything simpler ; compounds 					t be
				erent types of atom / elements (bonded) ;		[2]
		(ii)	covalent ; bonded eler	ments are both non-metals / compounds a	are gases ;	[2]
		 (iii) pure water is neutral ; rain water (generally) is (more) acidic (than pure water) ; rainwater during thunderstorm is the most acidic / owtte ; likely to be caused by nitrogen oxides ; 				
				reacted with the rain to form an acid ;		[max 2]
						[Total: 9]
9	 (a) nutrition excretion respiration reproduction growth movement (1 mark for any two correct);; 				[max 2]	
	(b)		adrenaline ;			[1]
	 (ii) increases pulse rate ; increases blood pressur increases glucose in blo 			lood pressure ;		[max 2]
	(c) in the blood / blood vessels ;					[1]
						[Total: 6]