## UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

**International General Certificate of Secondary Education** 

## MARK SCHEME for the May/June 2012 question paper for the guidance of teachers

## 0653 COMBINED SCIENCE

0653/63

Paper 6 (Alternative to Practical), maximum raw mark 60

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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	Page 2		Mark Scheme: Teachers' version	Syllabus	Paper
			IGCSE – May/June 2012	0653	63
1	(a) (i)	37 °0 body	C; rtemperature/optimal for enzymes/owtte;		[2]
	(ii)	205,	217, 185 ;;		[2]
	(iii)	(iii) 202 s; (allow 1 mark max in parts (i) and (ii) if times only given in minutes)		in minutes)	[1]
	(b) pink <u>due</u> to sodium carbonate; fat is digested/broken down; <u>fatty</u> acids neutralise the alkali;				Ol
	causing phenolphthalein to change colour/neutralise;				[max 2]
	(c) to ensure contents/tubes reach the temperature/all tubes the same temp/body temp;				
	(d) EITHER repeat with boiled/heated/denatured lipase (demonstrates it is an enzyme); no change in pink colour/no reaction/very long time to change colour; OR				
	•	repeat with different types of fat or named fat (demonstrates it breaks down fats reaction works as before/owtte;		[max 2]	
					[Total: 10]
2	(a) 13.7	7;			[1]
	(b) (i)	exte	th ( <i>I</i> ) = 7.8 ; rnal diameter, ( <b>d</b> <sub>e</sub> ) = 2.5 ; nal diameter, ( <b>d</b> <sub>i</sub> ) = 1.8 ;		[3]
	(ii)	$2.5^2$ = 3.0	– 1.8 <sup>2</sup> ; (allow ecf) 01 ;		[2]
	(iii) $-(V) = 3.14 \times 3.01 \times 7.8 \div 4 = $ ; (allow ecf)				
		(betv	ween) 18.1 and 18.5 ;		[2]
	(c) (formula used) density = mass/volume; 0.74; (allow ecf from incorrect values, but <b>not</b> from incorrect formula)				[2]
					[Total: 10]

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3 (a) 20.<u>0</u>; 47.5; 43.5; (no tolerance) [3]

(b) (i) axes correct and both labelled with units; points correctly plotted; smooth curve through points; maximum;

[4]

(iii) from graph (should be about 34 but accept 32);

[1]

(iii) substitution 25 × 4.2 × ans (b)(iii);

[2]

correctly worked out if use 34 = 3360;

[Total: 10]

(a) (i) correct answers in column 3;

[1]

time after drinking coffee/min	number of beats in 30 s	number of beats per min
0	36	72
5	39	78
10	42	84
15	45	90
20	45	90
25	37	74
30	36	72

(ii) suitable axes (scale and labels); plotting correct;

decent curve drawn;

[3]

(iii) correct estimate from graph (about 17.5); (do **not** allow range)

[1]

[1]

(b) (i) exercise causes heart rate to increase (therefore not a fair test);

(ii) volume of coffee;

concentration of coffee;

(amount of/quantity of coffee – max 1)

[2]

(iii) take readings more frequently (e.g. every 2 minutes); would see more clearly the peak in heart rate; more readings between 15 and 20 minutes;

[max 2]

[Total: 10]

		<b>.</b>	IGCSE – May/June 2012	0653	63
5 (	(a)	(i)	9 (cm);		[1]
		(ii)	9 × 30 = 270 ; × 2 = 540 (m) ;		[2]
			allow any sensible idea, e.g. distracted/forgot/not concerns correct sound owtte;	ar	
			(NOT just timing / experimental error)		[1]
	(	(iv)	1.76(5); (allow 1.76 or 1.77)		[1]
		(v)	using <u>their</u> value from above ÷ <u>their</u> distance ; answer ; e.g. 540 ÷ 1.765 = 306		[2]
		(vi)	must comment on their value, e.g. accurate as values inaccurate as values far apart;	are close together	r/ [1]
(b) any two of the following: longitudinal wave; (requires) molecules/particles; closer together;			ngitudinal wave ; equires) molecules/particles ;		[max 2] [ <b>Total: 10]</b>
					[101
6	(a)		<u>hted</u> splint ; ps/small explosion etc ;		[2]
	(b)	(i)	bubbles/gas/hydrogen floats Mg to surface/owtte;		[1]
		(ii)	(copper) doesn't react with acid;		[1]
(	(c)		agnesium + copper produces hydrogen faster/steeper grap pper acts as a catalyst/hydrogen given off faster (if say ste		[2]
(	(d)	son	me magnesium/solid remains ;		[1]
(	(e)		etch below others ; nd) reaches same level ;		[2]
	(f)	con	nnected to a syringe (labelled or graduations shown) ;		[1]
	-		· · · · · · · · · · · · · · · · · · ·		[Total: 10]

Mark Scheme: Teachers' version

Syllabus

Paper

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