

# MEASURING DENSITY

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1 A student is trying to find the density of water and of a large, regularly-shaped solid.

Which apparatus is needed to find the density of **both**?

- A balance, clock, ruler
  - B balance, measuring cylinder, ruler
  - C balance, measuring cylinder, string
  - D clock, ruler, string
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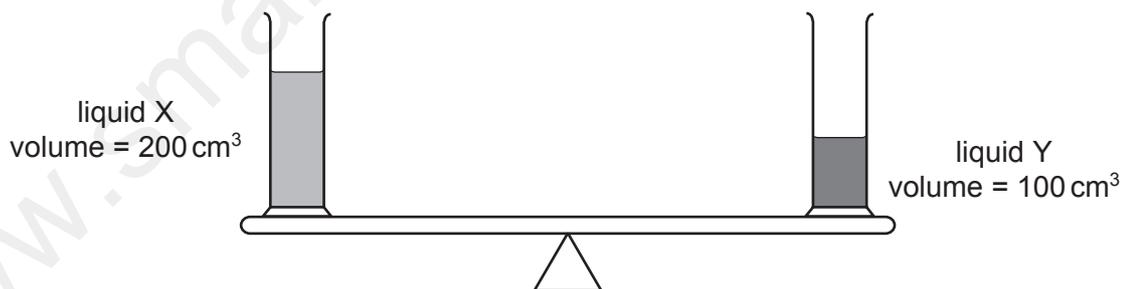
2 A metal drum has a mass of 200 kg when empty and 1000 kg when filled with 1.0 m<sup>3</sup> of methylated spirit.

What is the density of methylated spirit?

- A 0.0050 kg/m<sup>3</sup>
  - B 0.11 kg/m<sup>3</sup>
  - C 800 kg/m<sup>3</sup>
  - D 1000 kg/m<sup>3</sup>
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3 Two identical measuring cylinders containing different liquids are placed on a simple balance.

They balance as shown.



How does the density of X compare with the density of Y?

- A density of X =  $\frac{1}{2}$  × density of Y
  - B density of X = density of Y
  - C density of X = 2 × density of Y
  - D density of X = 4 × density of Y
-

4 A student needs to find the density of a cubic block of wood.

Which two pieces of apparatus should she use?

- A balance and metre rule
  - B balance and thermometer
  - C measuring cylinder and metre rule
  - D measuring cylinder and thermometer
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5 A student is trying to find the density of water and of a large, regularly shaped concrete block.

Which apparatus is needed to find the density of **both** the water and the concrete block?

- A balance, clock, measuring cylinder
  - B balance, clock, ruler
  - C balance, measuring cylinder, ruler
  - D clock, measuring cylinder, ruler
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6 A student is trying to find the density of water and of a large, regularly shaped concrete block.

Which apparatus is needed to find the density of **both** the water and the concrete block?

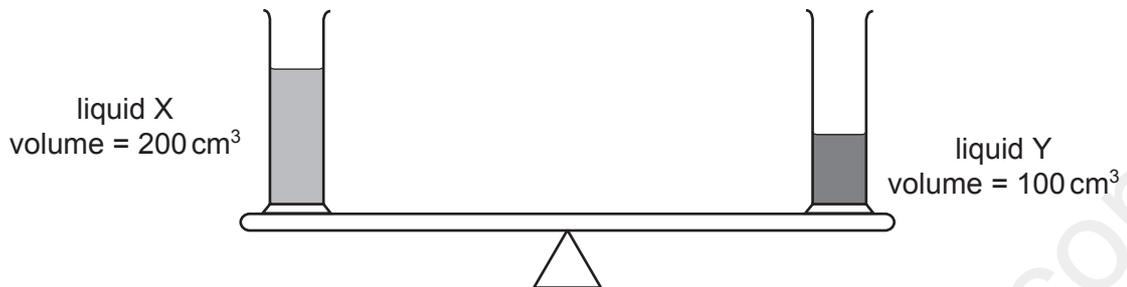
- A balance, clock, measuring cylinder
  - B balance, clock, ruler
  - C balance, measuring cylinder, ruler
  - D clock, measuring cylinder, ruler
- 

7 A student is told to measure the density of a liquid and also of a large cube of metal.

Which pieces of equipment are sufficient to be able to take the measurements needed?

- A balance, measuring cylinder and ruler
  - B balance and thermometer
  - C measuring cylinder and ruler
  - D measuring cylinder, ruler and thermometer
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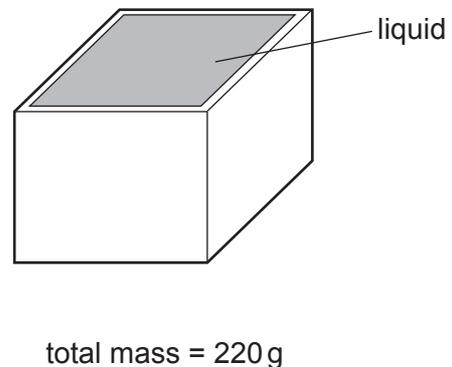
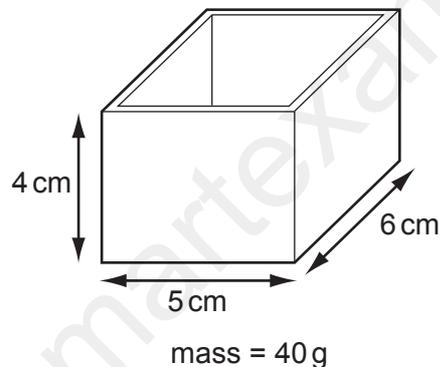
- 8 Two identical measuring cylinders containing different liquids are placed on a simple balance. They balance as shown.



How does the density of X compare with the density of Y?

- A density of X =  $\frac{1}{2}$  × density of Y  
 B density of X = density of Y  
 C density of X = 2 × density of Y  
 D density of X = 4 × density of Y

- 9 The diagrams show a rectangular box with inside measurements of 5 cm × 6 cm × 4 cm.



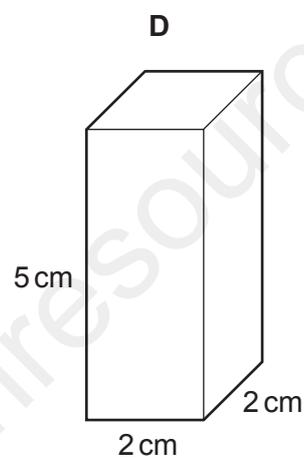
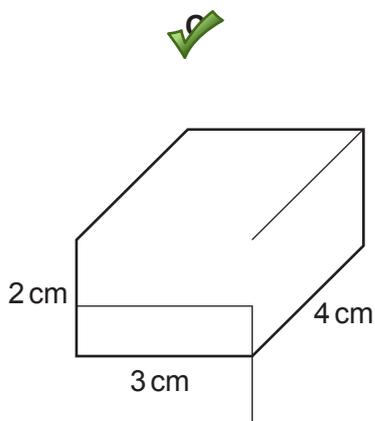
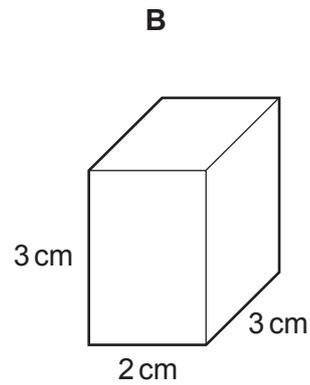
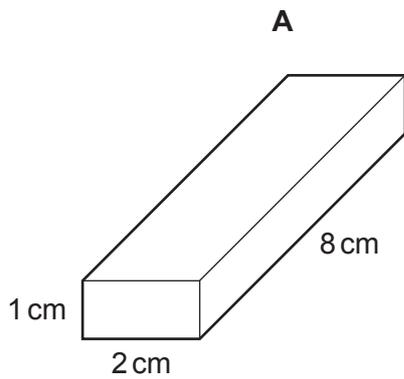
The box has a mass of 40 g when empty. When filled with a liquid, it has a total mass of 220 g.

What is the density of the liquid?

- A  $\frac{220}{(5 \times 6 \times 4)}$  g/cm<sup>3</sup>       C  $\frac{(5 \times 6 \times 4)}{220}$  g/cm<sup>3</sup>  
 B  $\frac{(220 - 40)}{(5 \times 6 \times 4)}$  g/cm<sup>3</sup>       D  $\frac{(5 \times 6 \times 4)}{(220 - 40)}$  g/cm<sup>3</sup>

**10** The diagrams show four blocks with the same mass.

Which block is made from the least dense material?

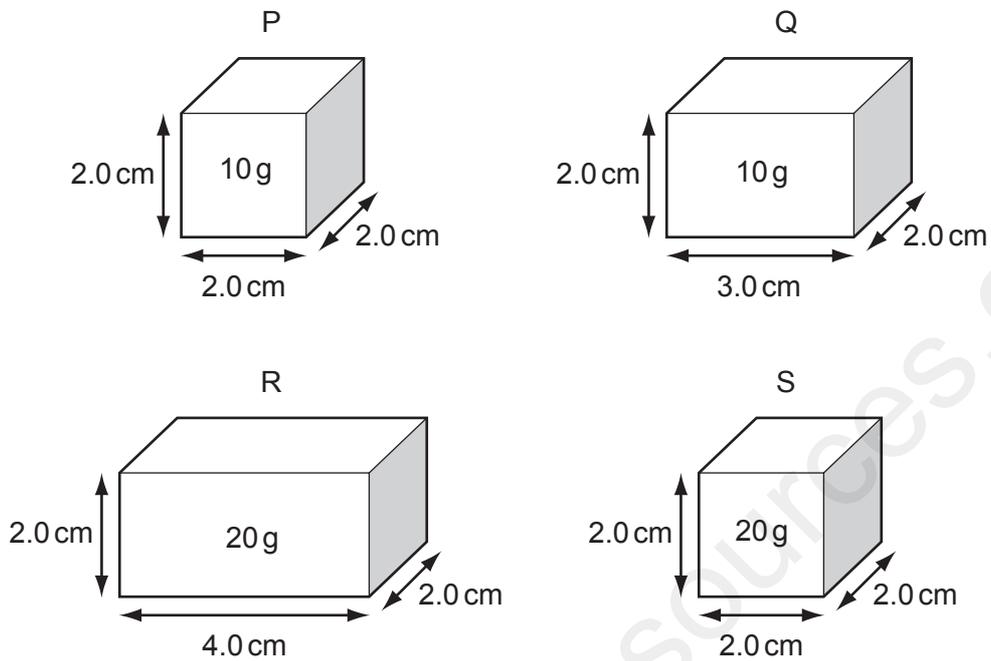


**11** A liquid has a density of  $0.80 \text{ g/cm}^3$ .

Which could be the volume and mass of this liquid?

	volume/cm <sup>3</sup>	mass/g
<b>A</b>	2.0	16
<b>B</b>	8.0	10
<b>C</b>	10	8.0
<b>D</b>	16	2.0

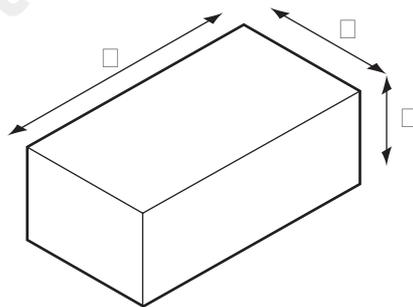
- 12 Four rectangular blocks, P, Q, R and S are shown. Each block is labelled with its size and its mass.



Which two blocks have the same density?

- A P and Q     B P and R    C Q and R    D R and S

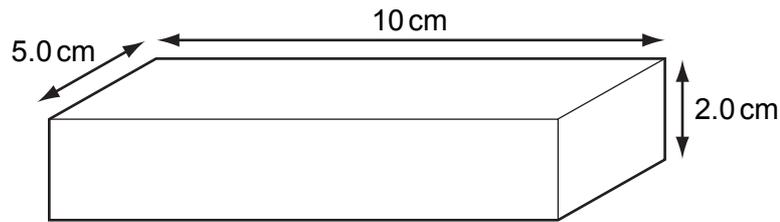
- 13 The diagram shows the dimensions of a rectangular block of metal of mass  $m$ .



Which expression is used to calculate the density of the metal?

- A  $m \times p \times q$   
 B  $m \times p \times q \times r$   
 C  $\frac{m}{(p \times q)}$   
 D  $\frac{m}{(p \times q \times r)}$

- 14 The diagram shows a rectangular metal block measuring  $10\text{ cm} \times 5.0\text{ cm} \times 2.0\text{ cm}$ .

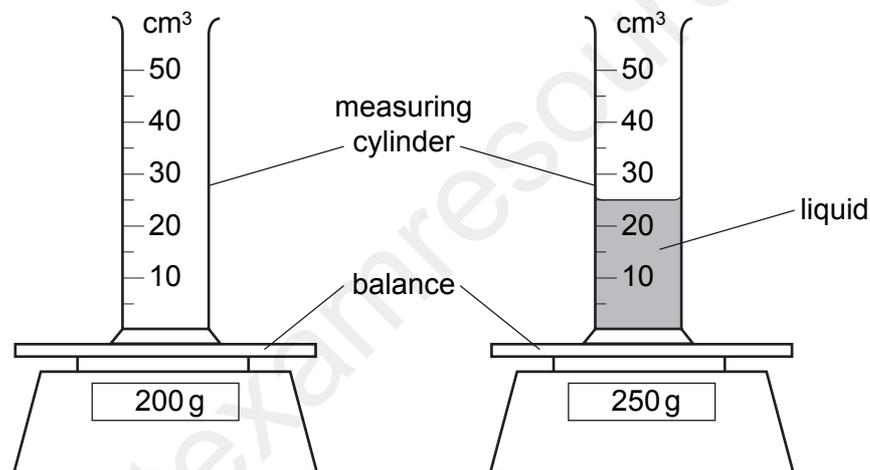


Its mass is 250 g.

What is the density of the metal?

- A  $0.20\text{ g/cm}^3$     B  $0.40\text{ g/cm}^3$      C  $2.5\text{ g/cm}^3$     D  $5.0\text{ g/cm}^3$
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- 15 The diagram shows an experiment to find the density of a liquid.



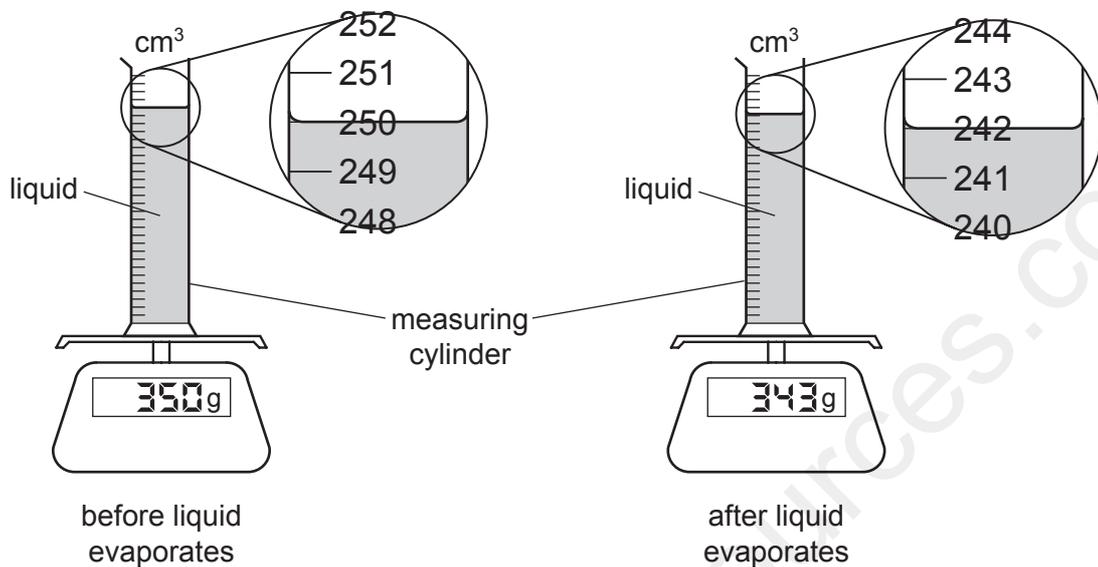
What is the density of the liquid?

- A  $0.5\text{ g/cm}^3$      B  $2.0\text{ g/cm}^3$     C  $8.0\text{ g/cm}^3$     D  $10.0\text{ g/cm}^3$
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- 16 Which items of apparatus are required to determine the density of a liquid?

- A balance and measuring cylinder  
B balance and thermometer  
C metre rule and measuring cylinder  
D metre rule and thermometer
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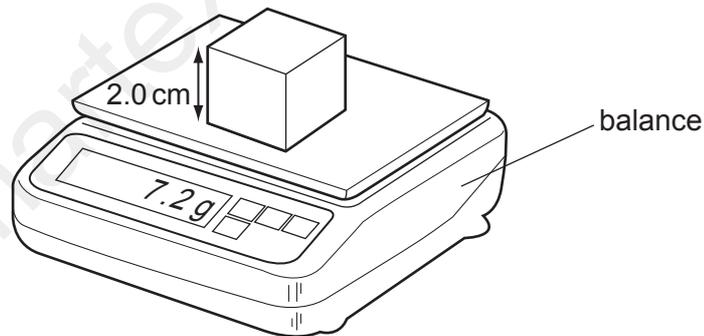
- 17 A measuring cylinder containing liquid is placed on a top-pan balance. The apparatus is left overnight and some of the liquid evaporates. The diagrams show the readings.



What is the density of the liquid?

- A 0.875 g/cm<sup>3</sup>    B 1.14 g/cm<sup>3</sup>    C 1.40 g/cm<sup>3</sup>    D 1.42 g/cm<sup>3</sup>
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- 18 A cube of side 2.0 cm is placed on a balance.



What is the density of the cube?

- A 0.90 g/cm<sup>3</sup>    B 1.2 g/cm<sup>3</sup>    C 1.8 g/cm<sup>3</sup>    D 3.6 g/cm<sup>3</sup>
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