

NO:	STRUCTURE OF AN ATOM-SET-1															
1	<p data-bbox="228 260 1425 291">Which row gives the number of electrons in the outer electron shell of fluorine and of neon?</p> <table border="1" data-bbox="228 327 591 611"> <thead> <tr> <th data-bbox="228 327 302 380"></th> <th data-bbox="302 327 448 380"><math>{}^{19}_{9}\text{F}</math></th> <th data-bbox="448 327 591 380"><math>{}^{20}_{10}\text{Ne}</math></th> </tr> </thead> <tbody> <tr> <td data-bbox="228 380 302 443"><b>A</b></td> <td data-bbox="302 380 448 443">7</td> <td data-bbox="448 380 591 443">8</td> </tr> <tr> <td data-bbox="228 443 302 506"><b>B</b></td> <td data-bbox="302 443 448 506">7</td> <td data-bbox="448 443 591 506">10</td> </tr> <tr> <td data-bbox="228 506 302 569"><b>C</b></td> <td data-bbox="302 506 448 569">9</td> <td data-bbox="448 506 591 569">8</td> </tr> <tr> <td data-bbox="228 569 302 611"><b>D</b></td> <td data-bbox="302 569 448 611">9</td> <td data-bbox="448 569 591 611">10</td> </tr> </tbody> </table>		${}^{19}_{9}\text{F}$	${}^{20}_{10}\text{Ne}$	<b>A</b>	7	8	<b>B</b>	7	10	<b>C</b>	9	8	<b>D</b>	9	10
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Ms-1	A															
2	<p data-bbox="228 772 1425 804">In the molecules <math>\text{CH}_4</math>, <math>\text{HCl}</math> and <math>\text{H}_2\text{O}</math>, which atoms use <b>all</b> of their outer shell electrons in bonding?</p> <p data-bbox="228 835 1133 867"><b>A</b> C and Cl      <b>B</b> C and H      <b>C</b> Cl and H      <b>D</b> H and O</p>															
Ms-2	B															

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