

NO:	STRUCTURE OF AN ATOM-SET-1															
1	<p data-bbox="228 260 1425 291">Which row gives the number of electrons in the outer electron shell of fluorine and of neon?</p> <table border="1" data-bbox="228 327 591 611"> <thead> <tr> <th data-bbox="228 327 302 380"></th> <th data-bbox="302 327 448 380">${}^{19}_{9}\text{F}$</th> <th data-bbox="448 327 591 380">${}^{20}_{10}\text{Ne}$</th> </tr> </thead> <tbody> <tr> <td data-bbox="228 380 302 443">A</td> <td data-bbox="302 380 448 443">7</td> <td data-bbox="448 380 591 443">8</td> </tr> <tr> <td data-bbox="228 443 302 506">B</td> <td data-bbox="302 443 448 506">7</td> <td data-bbox="448 443 591 506">10</td> </tr> <tr> <td data-bbox="228 506 302 569">C</td> <td data-bbox="302 506 448 569">9</td> <td data-bbox="448 506 591 569">8</td> </tr> <tr> <td data-bbox="228 569 302 611">D</td> <td data-bbox="302 569 448 611">9</td> <td data-bbox="448 569 591 611">10</td> </tr> </tbody> </table>		${}^{19}_{9}\text{F}$	${}^{20}_{10}\text{Ne}$	A	7	8	B	7	10	C	9	8	D	9	10
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A	7	8														
B	7	10														
C	9	8														
D	9	10														
2	<p data-bbox="228 722 1425 753">In the molecules CH_4, HCl and H_2O, which atoms use all of their outer shell electrons in bonding?</p> <p data-bbox="228 779 1133 810">A C and Cl B C and H C Cl and H D H and O</p>															